CH	EMICAL vs. PHYSICAL CHANGE	Name
In a physical change, the original substance still exists, it has only changed in form. Energy changes usually do not accompany physical changes, except in phase changes and when substances dissolve.		
In a chemical change, a new substance is produced. Energy changes always accompany chemical changes. Physical changes usually accompany chemical changes.		
Classify the following as being either a chemical or a physical change.		
1.	Sodium chloride dissolves in water.	
2.	Hydrochloric acid reacts with sodium hydroxide to produce a salt, water and heat.	
3.	A pellet of sodium is sliced in half.	
4.	Water is heated and changed to steam.	
5.	Food is digested.	
6.	Starch molecules are formed from smaller glucose molecules.	
7.	Ice melts.	
8.	Plant leaves lose water through evaporation.	
9.	A red blood cell placed in distilled water will swell and burst.	
10.	The energy in food molecules is transferred into molecules of ATP.	
11.	The roots of a plant absorb water.	
12.	Iron rusts.	
13.	Oxygen is incorporated into hemoglobin to bring it to the cells.	
14.	A person gets cooler by perspiring.	
15.	Proteins are made from amino acids.	
16.	A match burns.	
17.	A toothpick is broken in half.	
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ELEMENTS, COMPOUNDS AND MIXTURES

Name _____

An element consists of only one kind of atom. A compound consists of two or more different elements chemically combined in a fixed ratio. The components of a mixture can be in any proportion and are not chemically bound.

Classify each of the following as an element, compound or mixture by writing E, C or M in the space provided.

- 1. sodium _____
- 2. water _____
- 3. soil _____
- 4. coffee _____
- 5. oxygen _____
- 6. alcohol ____
- 7. carbon dioxide _____
- 8. cake batter _____
- 9. air ____
- 10. soap _____
- 11. iron _____
- 12. salt water _____
- 13. ice cream _____
- 14. nitrogen _____
- 15. eggs _____
- 16. blood _____
- 17. table salt _____
- 18. nail polish ____
- 19. milk _____
- 20. cola _____





